



# Assessment of Physiochemical Properties and Heavy Metals Concentration of Municipal Solid Compost (MSWC): A Case Study in Sokoto Metropolis, Nigeria

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## Abstract

Nowadays, farmers use municipal solid waste compost (MSWC) as fertilizer on their farmlands. Very scarce information on the quality of MSWC compost is available. As a result, the physiochemical characteristics of compost from ten (10) different dump sites in Sokoto City were evaluated in this study, and the results were compared with standards. According to the findings, the compost has a pH value ranging from 6.78 to 7.47, an electric conductivity between 2.16 to 6.48, organic carbon between (10.0% to 16.5%), organic matter between (0.94% to 14.78%), and a C:N ratio between 2.28 to 9.51. The heavy metals analysis shows an average value of  $275.47 \pm 89.86$ ,  $1.82 \pm 1.04$ ,  $82.44 \pm 28.28$ ,  $263.46 \pm 91.46$ ,  $45.23 \pm 11.72$ ,  $1.53 \pm 0.095$ ,  $71.71 \pm 21.46$  and  $0.29 \pm 0.39$  mg/kg for Zn, Cd, Fe, Cu, Ni, As, Pb, Mn and Cr respectively. The results of the analysis of variance revealed that the levels of heavy metals did not differ significantly between the locations (p-value 0.05). Some of the parameters studied were found to be within the permissible limits established by USA, UNBS/ICS, and WHO regulations.

## Subject Areas

Environmental Chemistry

## Keywords

Physiochemical, Municipal Solid Waste, Compost, Heavy Metal

## 1. Introduction

Municipal solid waste is a by-product of human activities, that is unwanted and

abandoned (Ibikunle *et al.*, 2018) [1]. These wastes are collected by waste collectors and vary by location (Lin *et al.*, 2015 [2]; Huseyin *et al.*, 2016 [3]; Read, 1999 [4]). Combustible and non-combustible materials make up the waste generated and their indiscriminate disposal leads to environmental degradation, pollution and global warming (Zhou *et al.*, 2014 [5]; Ibikunle *et al.*, 2018 [1]). According to the latest estimates from the World Bank (2022) [6], 1.3 billion tons of solid waste is generated globally every year, or 1.2 kg per person per day. The situation is expected to worsen as 2.2 billion tons of waste is expected to increase by 2025 (Iziorworu and Akpa, 2018) [7]. However, out of 32 million tons of solid waste Nigeria produced each year only 20-30% is recycled (Bakare, 2022) [8] and the remaining end in open dumpsite. According to Babatunde *et al.* (2013) [9], about 25 million tons of municipal solid wastes are generated annually, at 0.44 to 0.66 kg per day. This harms the environment but if managed will make adequate materials available for energy production, composting, and recycling, respectively (Bhat *et al.*, 2018) [10]. However, rapid urbanization in Nigeria has made municipal solid waste management a serious environmental issue (Chen *et al.*, 2012 [11]; Banna *et al.*, 2014 [12]; Ziraba *et al.*, 2016 [13]). In sub-Saharan African countries like Nigeria, waste management receives less than 10% of urban council budgets (Oosterveer *et al.*, 2010) [14]. This leads to uncollected MSW and poor management of landfills (Rogger *et al.*, 2011 [15]; Komakech *et al.*, 2014 [16]; Komakech *et al.*, 2016 [17]). Most of these wastes (more than 90%) are used for unscientific land filling or uncontrolled dumping on the outskirts of towns and cities (Sharholy *et al.*, 2008 [18]; Narayana, 2009 [19]). However, excessive waste in soil may increase the heavy metal content in groundwater and soil. Soil, crops and human health can all be negatively affected by heavy metals (Nyle and Ray 1999 [20]; Smith *et al.*, 1996 [21]).

Time is running out; our world will soon be covered in waste if waste is not managed properly (Nadeem, Farhan, and Ilyas, 2016) [22]. Composting of MSW appears to be an excellent choice for disposing of MSW. It's an inexpensive technology to keep organic waste out of landfills and produce quality products for agriculture (Nsimbe *et al.*, 2018) [23]. Municipal waste increased nitrogen, pH, cation exchange capacity, % alkali saturation and organic matter, it was claimed. Given that organic waste can enhance plant development by providing nutrients, continued land use of these wastes may be encouraged (Anikwe and Nwobodo, 2001 [24]; Nyles and Ray 1999 [20]). MSWC can be used in agriculture as a soil conditioner and fertilizer, which is more cost-effective for small farmers to use than inorganic fertilizers (Mandal *et al.*, 2014 [25]; Jodar *et al.*, 2017 [26]). In addition, there are concerns about the toxicity of solid waste when it is separated at source to reduce composting (Hargreaves *et al.*, 2008) [27].

Lack of pre-separation during composting can result in poor quality compost. Crop growth is negatively affected by nutrient abundance and deficiency, which also adversely affects the soil and creates bioaccumulation (Emamverdian *et al.*, 2015) [28]. Toxic metals migrate into crops as a result of application of municipal solid waste compost to agricultural fields in excess of heavy metals. Large

amounts of these heavy metals can cause nausea, vomiting, diarrhea, and pneumonia (Jaishankar *et al.*, 2014) [29]. These heavy metals can bind to biological components. Depending on the characteristics of the soil and compost, it is crucial to evaluate the characteristics and heavy metal content of the compost before applying it as a soil supplement in agricultural areas.

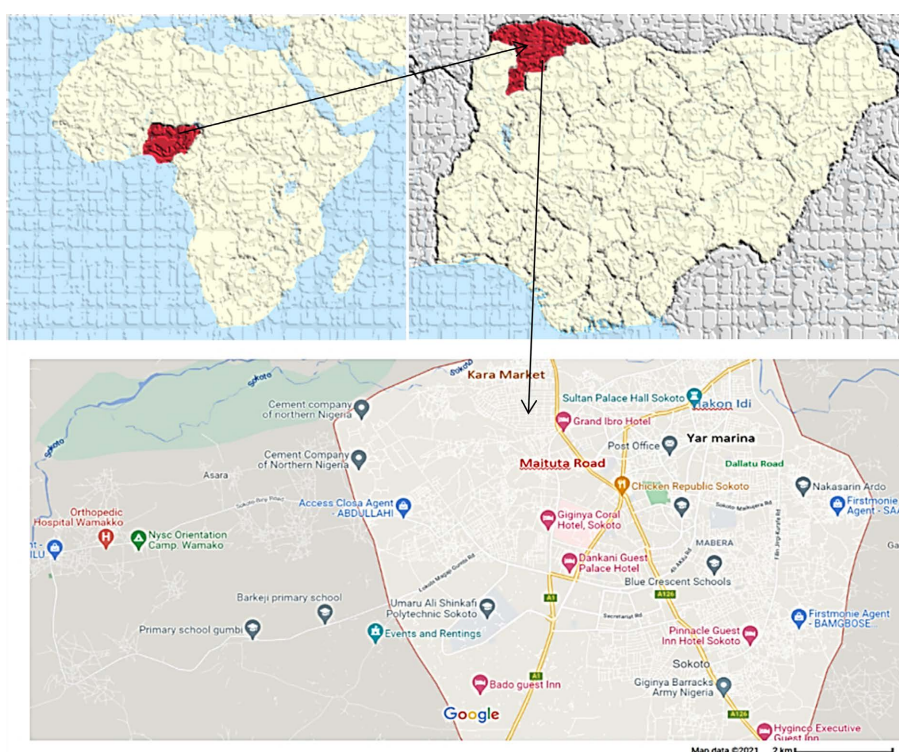
This study, therefore aimed at assessing physiochemical parameters and concentration of some selected heavy metals in solid municipal waste compost from different dumpsites within Sokoto metropolis.

### 1.1. Study Area

Sokoto is a city located in the extreme northwest of Nigeria, near the confluence of the Sokoto River and the Rima River (**Figure 1**). It occupies 25,973 square kilometres with a population of 563,861. Geographically situated between latitude  $130^{\circ}.05$  and  $13^{\circ}.0830$  north and longitude  $05^{\circ}15$  and  $5.250^{\circ}$  east with an average elevation of 272 m above sea level. The ten (10) sampling dumpsites selected were all within Sokoto Metropolis. These were Bado (BDO), Dallatu (DLT), Gawon-Nama (GNM), Kalanbaina (KBN), Kwannawa (KMW), Gidan dare (GDR), Marina (MRN), Tudun-wada (TWD) and Runbukawa (RKW).

### 1.2. Sample Collection and Treatment

Compost samples were collected in triplicate from various windrow depths and portions at each location (compost plants) to create composite samples. To guarantee complete population representation, samples of the compost manure



**Figure 1.** Location of the dumpsites area studied in Sokoto, Nigeria.

were taken from the upper 2 - 3 m, middle 1 - 2 m, and lower 0 - 1 m levels. Each sample weighed about 200 g. Each composite sample was thoroughly mixed, separated, and 20 g of it was sub-sampled (reduced). This portion was then air-dried at room temperature to stop biological activity, ground in a mechanical motor and pestle, sieved through an agriculture 2 mm screen to ensure a homogeneous mixture, placed in air-tight, labelled, clean polyethylene bags, transported to the lab in an icebox, and stored for additional analysis.

## 2. Materials and Methods

### 2.1. Determination of Total P and Total N

Total P and Total N were determined by Kjeldhal process using a standard AOAC Method 978.02, as reported by Okalebo *et al.*, (2002) [30].

### 2.2. Determination of Ca, Na, Mg and K

The samples were digested using wet digester method. After digestion, the produced solutions were used to analyzed K and Na using flame photometer by inserting appropriate filter. Ca and Mg content of the samples were determined using EDTA titration method (Horwitz *et al.*, 2005) [31] and their amount calculated using Equations (1) & (2)

$$\text{Ca}^{2+}\text{Mg}^{2+} = \frac{T \times N \times V}{\text{Mils of Aliquot} \times wt} \times 100 \quad (1)$$

$$\text{Ca}^{2+} = \frac{T \times N \times V}{\text{Mils of Aliquot} \times wt} \times 100 \quad (2)$$

where:  $T$  is titre of the sample,  $N$  normality of EDTA,  $wt$  weight of sample and  $V$  of the leachate collected.

Amount of  $\text{Mg}^{2+}$  was determined by subtracting amount of  $\text{Ca}^{2+}$  from  $\text{Ca}^{2+}\text{Mg}^{2+}$  obtained in Equation (1)

### 2.3. Proximate Analysis

The pH and EC of the samples were measured in aqueous suspension using a pH meter and conductivity meter in 1:10 % using AOAC standard method (AOAC 973.04) [32]

The moisture contents of the samples were determined using standard test method for residual moisture analysis (ASTM 3173) [33].

$$\% \text{ Moisture Content} = \frac{W_2 - W_1}{W_2 - W_0} \times 100 \quad (3)$$

where  $W_0$  = weight of empty crucible,  $W_2$  = weight of crucible and the sample before oven drying, and  $W_1$  = weight of crucible and the sample after drying.

Ash content of the samples was determined using Equation (4), following a standard ASTM method (ASTMD, 3174) [34]

$$\text{Ash} (\%) = \frac{W_1 - W_0}{W_2 - W_0} \times 100 \quad (4)$$

where  $W_0$  = weight of empty crucible,  $W_2$  = weight of crucible and the sample before combustion and  $W_1$  = weight crucible and the sample after combustion.

Organic Volatile Matter was determined using standard method (ASTMD, 3175) [35] and calculated using Equation (5)

$$V(\%) = \frac{W_1 - W_2}{W_1} \times 100 \quad (5)$$

where  $W_1$  = weight of the sample and  $W_2$  = weight of the Sample after incineration

Fixed carbon was determined by difference using Equation (6).

$$\text{Fixed Carbon (\%)} = 100 - (\% \text{ moisture content} + \% \text{ Ash} + \% \text{ volatile matter}) \quad (6)$$

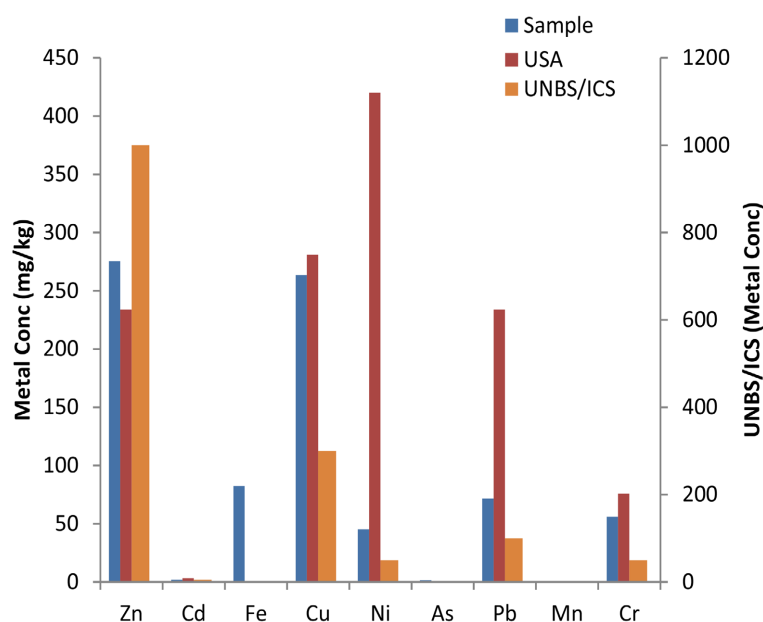
## 2.4. Heavy Metal Analysis

Heavy metals were determined by atomic absorption spectroscopy (AAS) using the standard methods adapted by USEPA 7000a (Schrenk *et al.*, 2012) [36]. Extractor and AAS controls were turned on and the thin tube, atomizer parts was cleaned with purge wire, while the burner opening with an alignment card. The ASS worksheet was opened. The lamp was turned on, to allow the light beam from the cathode strike the target area of the card. The fine was placed in 10 ml graduated chamber filled with deionized water. An analytical blank was prepared together with series of calibration solutions of known amounts of analyte. Blanks and standards were nebulized in sequence and their responses measured. A calibration chart is drawn for each solution, after which the sample solution is nebulized and measured. The metal concentration was determined based on the absorbance obtained for the unknown sample from the calibration.

## 3. Results and Discussion

One of the parameters used to check the quality of compost is its heavy metal concentration. Dyes, batteries, electronics, cosmetics and pharmaceutical residues are the main sources of heavy metals in compost (Amouei *et al.*, 2009) [37]. Due to the harmful effects of heavy metals on humans and the environment, some standards have been established for the concentration of these metals in compost. In this study, the average concentrations of heavy metals detected in the compost from different landfills were  $275.47 \pm 89.86$ ,  $1.82 \pm 1.04$ ,  $82.44 \pm 28.28$ ,  $263.46 \pm 91.46$ ,  $45.23 \pm 11.72$ ,  $1.53 \pm 0.095$ ,  $71.71 \pm 21.46$  and  $0.29 \pm 0.39$  mg/kg for Zn, Cd, Fe, Cu, Ni, As, Pb, Mn and Cr respectively. The mean concentrations of the heavy metals were all within the allowed limits of the USA and the UNBS/ICS Standard (Figure 2). High concentrations of heavy metals limit the use of compost on farmland. Environmental impact of heavy metal-contaminated compost varies with soil type, plant species, and compost quality (Zhao *et al.* 2011) [38].

All the compost samples show high concentration of Zn compared to others heavy metals detected with Cd concentration been the least. This is in line with the findings of Ibrahim *et al.*, (2020) [39], for the heavy metals analysed in municipal solid waste from Potiskum (Yobe State, Nigeria). The higher concentration



**Figure 2.** Heavy metals concentration of the compost compared with the standard.

of Zn in the compost may be due to the stability of ZnO, as ZnO in soil has a higher stability coefficient (Ma and Rao, 1997) [40]. The low concentration of Cd may be attributed to the weak adsorption properties of Cd (Mido and Satake, 2003) [41].

### 3.1. Correlation between Heavy Metal Concentrations in the Compost

Pearson's correlation analysis was carried out to analyze correlations between heavy metal concentrations in the compost from different sites. Result of the analysis is shown in (Table 1). It represents the correlation between different dumps in terms of heavy metal concentrations. The analysis revealed a strong positive correlation of heavy metal between the sites. This may be due to the correlation in composition of the solid waste from the different dumping sites.

### 3.2. Statistical Analysis

A one-way analysis of variance (Table 2) was performed to determine significant differences in terms of heavy metal concentrations among the different studied sites. The analysis was based on probability level ( $\alpha$ ) of 0.05. The ANOVA results revealed that there were no significant differences in concentrations of heavy metals from the dump sites ( $p$ -value > 0.05). A low value of calculated F (0.2531) also predicts a non-significant difference when compared to the critical F value of 1.999.

The moisture content of the sample was in the range of 1.13 - 2.23 with an average value of  $2.00\% \pm 0.52\%$ . High moisture content reduces air space, so the compost can become clumpy. Dusty compost is often generated by compost ventilation, causing ambient air pollution in the surrounding area (Mandal *et al.*,

**Table 1.** Pearson's correlation coefficient for heavy metal concentrations in the compost.

	<i>BDO</i>	<i>DLT</i>	<i>FKI</i>	<i>GNM</i>	<i>GDR</i>	<i>KBN</i>	<i>KNW</i>	<i>MRN</i>	<i>TWD</i>	<i>RKW</i>
<i>BDO</i>	1									
<i>DLT</i>	0.6399	1								
<i>FKI</i>	0.8239	0.9140	1							
<i>GNM</i>	0.8496	0.7952	0.9691	1						
<i>GDR</i>	0.8407	0.8870	0.9942	0.9824	1					
<i>KBN</i>	0.8635	0.7934	0.9666	0.9981	0.9823	1				
<i>KNW</i>	0.9139	0.8278	0.9730	0.9754	0.9798	0.9823	1			
<i>MRN</i>	0.8804	0.8335	0.9799	0.9912	0.9889	0.9950	0.9936	1		
<i>TWD</i>	0.8747	0.8781	0.9840	0.9632	0.9842	0.9704	0.9932	0.9879	1	
<i>RKW</i>	0.7808	0.8719	0.9729	0.9590	0.9697	0.9637	0.9638	0.9752	0.9797	1

**Table 2.** Result of one way analysis of variance (ANOVA).

Source of Variation	SS	df	MS	F	P-value	F crit
Between Groups	29,948.41	9	3327.601	0.253112	0.984809	1.999115
Within Groups	1,051,739	80	13146.73			
Total	1,081,687	89				

2014) [25]. The pH of the compost varied from 6.78 to 7.47 with an average value of  $7.06 \pm 0.22$ , which indicates that it was almost acidic to slightly alkaline, and was found to be within the standard recommended range of 6 to 8 (WHO). A high pH accelerates the release of ammonia from the compost into the ambient air. A gradual increase in ammonia emissions to ambient air corresponds to an increase in compost temperature ( $35^{\circ}\text{C}$  -  $60^{\circ}\text{C}$ ) and alkaline pH value (Om-rani *et al.* 2004) [42]. Disease-causing pathogens may be present in compost, during waste sorting and transportation; pathogenic bacteria can cause illness in those who handle them. The values of pH obtained indicated that the composts were not adulterated with pathogenic bacteria. This may be due to increased temperatures, which destroy pathogens (Day and Shaw 2001) [43].

The nutrients in the compost were analyzed to determine its suitability for agricultural land. Nutrients are needed in compost piles because microbes need them to grow, besides their importance in terms of fertilizer value (Sadeghi *et al.*, 2015) [44]. Among the nutrients, Carbon, Nitrogen, Phosphorus, Potassium, Calcium, Iron, and Manganese are the most important (Abbasi *et al.*, 2010) [45]. The average values of Nitrogen, Phosphorus, and Potassium in the compost are  $2.29 \pm 0.86$ ,  $1.92 \pm 0.3$ , and  $55.27 \pm 13.68$  mg/kg, respectively. WHO standards for nitrogen, phosphorus and potassium in compost are 0.4 - 3.5, 0.3 - 3.8 and 0.5 - 1.8, respectively (Table 3). The values recorded in this study meet WHO standards with exception of potassium, which was found to be higher than the

**Table 3.** Physiochemical properties of the compost.

Parameters	Units	Range	Average	WHO Standard
pH		6.78 - 7.47	7.06 ± 0.22	6 - 8
Electrical conductivity (EC)	ds/m	2.16 - 6.48	3.31 ± 1.18	-
Organic carbon (O.C)	%	10.0 - 16.55	14.39 ± 1.94	-
Moisture content (M.C)	%	1.13 - 2.23	2.00 ± 0.52	-
Ash content (A.C)	%	87.55 - 95.53	92.54 ± 2.50	-
Organic matter (O.M)	%	0.94 - 14.78	7.02 ± 3.92	10% - 20%
Carbon/Nitrogen ratio (C:N)		2.28 - 9.51	7.02 ± 1.92	-
P	mg/kg	1.73 - 2.81	1.92 ± 0.3	0.3 - 3.8
N	mg/kg	1.52 - 4.48	2.29 ± 0.86	0.4 - 3.5
K	mg/kg	40.0 - 70.5	55.27 ± 13.68	0.5 - 1.8
Ca	mg/kg	1.0 - 2.50	1.49 ± 0.41	-
Na	mg/kg	80.0 - 207	155.24 ± 46.34	-
Mg	mg/kg	1.50 - 2.25	1.97 ± 0.27	-

specified limit. Electrical conductivity indicates concentration of dissolved salts in the compost. Composts with high concentrations of EC can cause problems such as soil salinity and biotoxicity (Malakootian *et al.*, 2014) [46]. The average electric conductivity of the compost was found to be  $3.31 \pm 1.18$ . High electrical conductivity in compost inhibits seed germination and plant growth (Rahman *et al.*, 2020) [47]. The C:N ratio of the composts varied from 2.28 to 9.51 with an average value of  $7.02 \pm 1.92$ . Stable mature composts that do not contain ligno-cellulosic material have CN ratios below 17 (Silva *et al.*, 2007) [48]. Some literature suggests an ideal value of 12, while others suggest a CN ratio of 20-40 as most suitable for use as a fertilizer (Rawat *et al.*, 2013) [49]. Composts with CN ratio values above or below the recommended values may inhibit seed germination, reduce plant growth, and cause phytotoxicity due to insufficient biodegradable organic matter, thereby damaging crops (Kabasiita *et al.*, 2022) [50]. Organic matters affect the quality of compost. The results in this study showed that organic matter in the compost ranges from 0.94 - 14.78 having average value of  $14.39\% \pm 1.94\%$ . Organic matter indicates the presence of heavy metals in compost as they are known to form complexes with organic matter (Jimoh and Sabo, 2013) [51]. A low OC indicates mature and stable composts (Kabasiita *et al.*, 2022) [50]; thus, these composts will be stable and usable for on-farm applications due to the low organic matter content of the composts.

#### 4. Conclusion

The application of municipal solid waste compost for agricultural purposes is rapidly growing. The composts are made of different sources of materials. To ensure the quality and safety of compost, it is necessary to analyze and monitor

its physical and chemical characteristics. In this study, the physiochemical properties and heavy metal analysis of municipal solid waste compost from different dumpsites within Sokoto City were investigated. Results indicated a good quality of the studied compost with only a limited number of parameters outside the specified limits of USA, UNBS/ICS, and WHO standards.

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### Conflicts of Interest

The authors declare no conflicts of interest.

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